

MOLECULAR MODELLING OF THE MOUSE CYTOCHROME P450 CYP2F2 BASED ON THE CYP102 CRYSTAL STRUCTURE TEMPLATE AND SELECTIVE CYP2F2 SUBSTRATE INTERACTIONS

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SUMMARY

The results of homology modelling of the mouse cytochrome P450, CYP2F2, are reported, based on the CYP102 crystallographic template. It is found that selective CYP2F2 substrates are able to fit the putative active site of the enzyme via aromatic π - π stacking and, in some cases, hydrogen-bonded interactions. Two alkyl-naphthalenes were investigated via the model to evaluate whether they are likely to act as CYP2F2 substrates and, of these, 2-isopropyl-naphthalene was found to fit the putative active site, whereas 2-(2-hexadecyl)naphthalene was unable to do this, due to its bulky side-chain. Consequently, it is possible to utilize homology modelling to evaluate the likelihood of enzyme-substrate selectivity for CYP2F2 and predict routes of metabolism mediated by this enzyme. This procedure can therefore be used to investigate the potential for activation of xenobiotics via this enzyme, especially those related to known CYP2F substrates, such as naphthalene.

KEY WORDS

cytochrome P450, xenobiotic metabolism and activation, mouse CYP2F2, modelling, CYP102 structural template

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INTRODUCTION

Enzymes of the cytochrome P450 (CYP) superfamily are associated with the mono-oxygenation of a large range and variety of organic compounds, both endogenous and exogenous /1,2/. Included in the latter category are pharmaceutical agents and environmental pollutants, many of which are destined for human exposure. Molecular modelling of the enzyme-substrate interactions for P450-mediated pathways can assist in the rationalization of substrate selectivity and site of metabolism, although this process involves initial construction of the enzymes themselves. One of the most appropriate crystal structure templates for homology modelling of mammalian microsomal P450s is represented by the CYP102 haemoprotein domain, which is available as the substrate-bound enzyme for the production of P450 models. There is interest in the role of CYP2F subfamily enzymes /3-5/ in the activation and metabolism of xenobiotics, some of which are likely to show toxicity (e.g. naphthalene) following metabolic activation via the mediation of this enzyme /6-11/. Consequently, it is anticipated that homology modelling of the mouse orthologue, CYP2F2, may yield important structural information regarding substrate preferences for this isoform, and also assist in evaluating routes of metabolism mediated by CYP2F enzymes in general. Although the mouse orthologue was modelled in this study, it is likely that the active sites of CYP2F enzymes in other mammalian species, including *Homo sapiens*, will be similar, due to the fact that there is a relatively high degree of homology between their protein sequences. Furthermore, it would appear that both mouse and human CYP2F enzymes possess similar substrate selectivities.

METHODS

Using the CYP102 haemoprotein domain crystal structure (pdb code: lfag) as a template the mouse CYP2F isoform, CYP2F2, was generated via amino acid sequence homology. Figure 1 provides a multiple sequence alignment of CYP102 against 20 CYP2 family enzymes to indicate similarities and differences between various sequences in this family, and this was employed in homology modelling of CYP2F2 via a procedure involving amino acid residue replacement, deletion and insertion as required, followed by energy

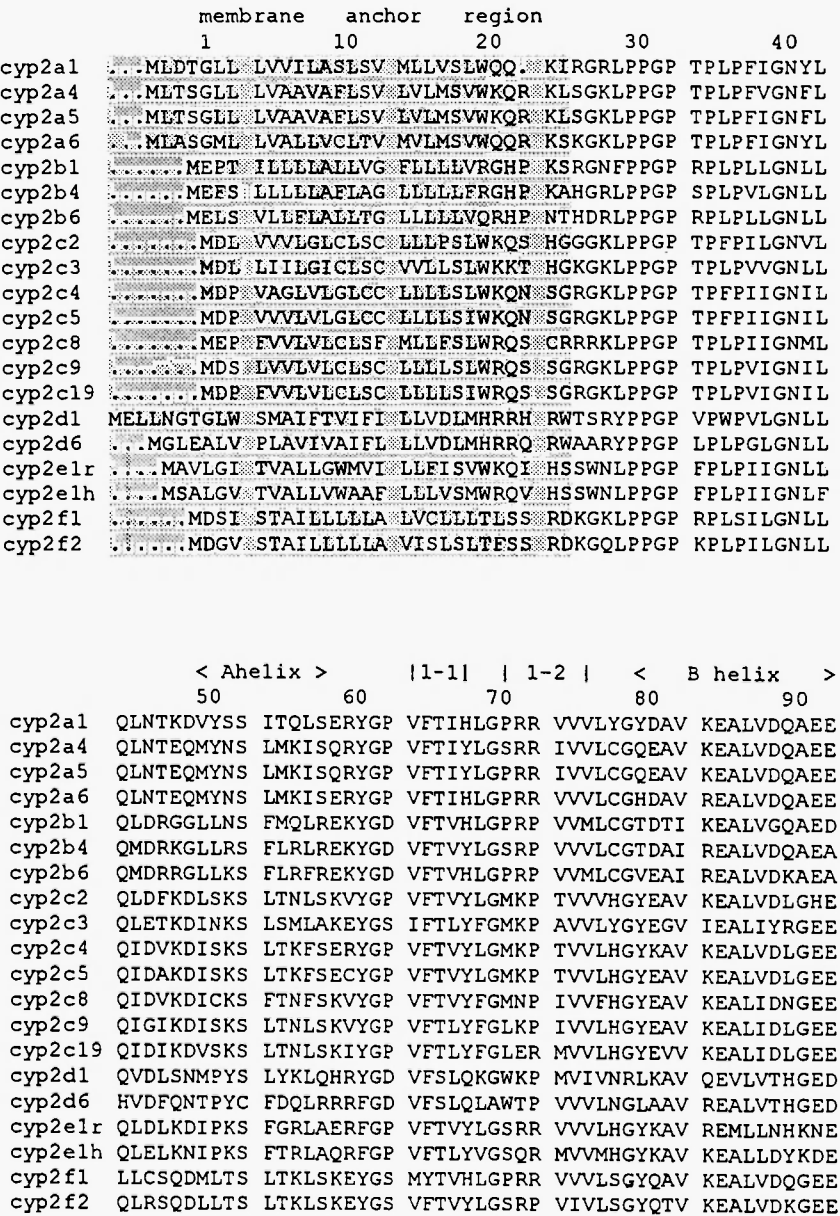


Fig. 1: A multiple sequence alignment of CYP102 and various CYP2 family proteins. The substrate recognition sites (SRS) are shaded and site-directed mutagenesis information is provided by emboldening the relevant residues.

	SRS1		< C helix >	
	100	110	120	130
cyp2a1	FSGRGEQATY	NTLEFKG...Y GVAFSS.GER	AKQLRRLSIA	TLRDFGVGKR
cyp2a4	FSGRGEQATE	DWLEFKG...Y GIAESS.GER	AKQLRSFSIA	TLRDFGVGKR
cyp2a5	FSGRGEQATF	DWLEFKG...Y GVVFSS.GER	AKQLRRFSIA	TLRDFGVGKR
cyp2a6	FSGRGEQATE	DWVLEFKG...Y GVVFSS.GER	AKQLRRFSIA	TLRDFGVGKR
cyp2b1	FSGRGETIAVI	EPIFHE...Y GVIFAN.GER	WKALRRFSIA	TMRDFGMGKR
cyp2b4	FSGRGKIANV	DPIFOG...Y GVIFAN.GER	WRALRRFSIA	TMRDFGMGKR
cyp2b6	FSGRGKIANV	DPIFRG...Y GVIFAN.GNR	WKVLRFSVT	TMRDFGMGKR
cyp2c2	LSGRSREIVT	AKLNKG...F GVIFSN.GKR	WTETRRFSIM	TLRNFGMGKR
cyp2c3	FSGRGIEPVE	DRVTKG...L GIVFSS.GEK	WKETRRFSLT	VLRNLGMGKK
cyp2c4	FAGRGHEPIA	EKVSKG...L GIVFTN.ANT	WKEMRRFSIM	TLRNFGMGKR
cyp2c5	FAGTGSVPIL	EKVSKG...L GIAFSN.AKT	WKEMRRFSIM	TLRNFGMGKR
cyp2c8	FSGRGNSPIS	QRITKG...L GIISN.GKR	WKEIRRFSLT	TLRNFGMGKR
cyp2c9	FSGRGIEPLA	ERANRG...F GIVFSN.GKK	WKEIRRFSLM	TLRNFGMGKR
cyp2c19	FSGRGHEPLA	ERANRG...F GIVFSN.GKR	WKEIRRFSLM	TLRNFGMGKR
cyp2d1	TADRPVPPIF	KCLGVKPRSQ	GVILASYGPE	WREQRFSVS
cyp2d6	TADRPVPPIF	QILGPGPRSQ	GVFLARYGPA	WREQRFSVS
cyp2e1r	FSGRGEIPAF	.REFKD...K GIIFNN.GPT	WKDTRRFSLT	TLRDYGMGKQ
cyp2e1h	FSGRGDLPAF	.HAYRD...F GIIFNN.GPT	WKDIRRFSLT	TLRNYGMGKQ
cyp2f1	FSGRGDYPAF	FNFTKG...N GIAESS.GDR	WKVLRQFSIQ	ILRNFGMGKR
cyp2f2	FSGRGAYPVE	FNFTRG...N GIAESS.GER	WKILRRFSVQ	ILRNFGMGKR

	< D helix >		< D' >		< Ehelix >
	140	150	160	170	180
cyp2a1	GVEERILEEA	GYLIKMLQGT	CGAPIDPTIY	LSKTVSNNVIS	SIVFGERFDY
cyp2a4	GIEERIQEEA	GFLIDSFRKT	NGAFIDPTFY	LSRTVSNNVIS	SIVFGDRFDY
cyp2a5	GIEERIQEEA	GFLIDSFRKT	NGAFIDPTFY	LSRTVSNNVIS	SIVFGDRFDY
cyp2a6	GIEERIQEEA	GFLIDAHRTG	GGANIDPTFF	LSRTVSNNVIS	SIVFGDRFDY
cyp2b1	SVEERIQEEA	QCLVEELRKS	QGAPLDPTFL	FQCITANIIC	SIVFGERFDY
cyp2b4	SVEERIQEEA	RCLVEELRKS	KGALLDNTLL	FHSITSNIIC	SIVFGKRFDY
cyp2b6	SVEERIQEEA	QCLIEELRKS	KGALMDPTFL	FQSITANIIC	SIVFGKRFHY
cyp2c2	SIEERVQEEA	HCLVEELRKT	NASPCDPTFI	LGAAPCNVIC	SVIFQNRFDY
cyp2c3	TIEERIQEEA	LCLIQALRKT	NASPCDPTFL	LFCVPCNVIC	SVIFQNRFDY
cyp2c4	SIEDRVQEEA	RCLVEELRKT	NALPCDPTFI	LGCAPCNVIC	SVILHNRFDY
cyp2c5	SIEDRIQEEA	RCLVEELRKT	NASPCDPTFI	LGCAPCNVIC	SVIFHNRFDY
cyp2c8	SIEDRVQEEA	HCLVEELRKT	KASPCDPTFI	LGCAPCNVIC	SVVFQKRFDY
cyp2c9	SIEDRVQEEA	RCLVEELRKT	KASPCDPTFI	LGCAPCNVIC	SIIFLKRFDY
cyp2c19	SIEDRVQEEA	RCLVEELRKT	KASPCDPTFI	LGCAPCNVIC	SIIFQKRFDY
cyp2d1	SLEEVVTKEA	GHLCDATAQ	AGQSINPKAM	LNKALCNVIA	SLIFARRFEY
cyp2d6	SLEQWVTEEA	ACLCAAFANH	SGRPFPRNGL	LDKAVSNVIA	SLTCGRRFY
cyp2e1r	GNEDRIQKEA	HFLLEELRKT	QGQPFDPFTFV	IGCTPFNVIA	KILFNRDFDY
cyp2e1h	GNESRIQREA	HFLLEALRKT	QGQPFDPFTFV	IGCAPCNVIA	DILFRKHFDY
cyp2f1	SIEERILEEG	SFLLEADVRKT	EGEPFDPFTFV	LSRSVSNIIC	SVLFGSRFDY
cyp2f2	SIEERILEEG	SFLLEVLKRM	EGKPFDPVFI	LSRSVSNIIC	SVVFGSRFDY

Fig. 1 cont.

	< F helix <i>SRS2</i> >		< F' >		< <i>SRS3</i>
	190	200	210	220	230
cyp2a1	EDTEFLSLLQ	MMGQNNRFAA	SPTGQLYDMF	HSVVMKYLPGP	QQQIIKVTQK
cyp2a4	EDKEFLSLLR	MMLGSLQETA	TSMGQVYEMF	SSVMKHLPGP	QQQAFKELQG
cyp2a5	EDKEFLSLLR	MMLGSLQETA	TSMGQLYEMF	SSVMKHLPGP	QQQAFKELQG
cyp2a6	KDKEFLSLLR	MMLGIFQETS	TSTGQLYEMF	SSVMKHLPGP	QQQAFQQLQG
cyp2b1	TDRQFLRLLE	LEYRTESLLS	SFSSQVFEFF	SGFLKYFPGA	HRQISKNLQE
cyp2b4	KDPVFLRLLD	LEFQSESLIS	SFSSQVFELF	PGFLKHFPPT	HRQIYRNLOE
cyp2b6	QDQEFKLMLN	LEFYTESLIS	SVFGQLFELF	SGFLKYFPGA	HRQVYKNLOE
cyp2c2	TDQDFLSLMG	KENENFKILN	SPWVQFCNCF	PILFDYFPGS	HRKAVKNIFY
cyp2c3	DDEKFKTLIK	YFHENFELLG	TPWIQLYNIF	PILGHYLPFS	HRQLFKNLIDG
cyp2c4	KDEEFLKLME	RINENFRLIS	SPWLQVYNNF	PALLDYFPGI	HKTLLKNADY
cyp2c5	KDEEFLKLME	SINENVRILS	SPWLQVYNNF	PALLDYFPGI	HKTLLKNADY
cyp2c8	KDQNFLLTMK	RNENEFRLIN	SPWQVCNCF	PLLIDCFPGT	HNKVLKNVAL
cyp2c9	KDQQLNLME	KLNENIKILS	SPWQVCNCF	SPIIDYFPGT	HNKLLKNVAF
cyp2c19	KDQQLNLME	KLNENIRIVS	TPWQVCNCF	PTIIDYFPGT	HNKLLKNLAF
cyp2d1	EDPYLIRMVK	LVEESLTVS	GFIPEVLNTF	PALLR.IPGL	ADKVFQOKT
cyp2d6	DDPRFLRLLD	LAQEGKKEES	GFLREVLNAV	PVLLH.IPAL	AGKVLRFQKA
cyp2e1r	KDKQALRLMS	LENENFYLLS	TPWLQVYNNF	SNYLQYMPGS	HRKVKIKNVSE
cyp2e1h	NDEKFLRLMY	LENENFHILS	TPWLQLYNNF	PSFLHYLPFS	HRKVKIKNVAF
cyp2f1	DDRLLTIIR	LINDNFQIMS	SPWGELYDIF	PSLLDWVPGP	HQRIFONFKK
cyp2f2	DDRLLTIIR	FINDNFQIMS	SPWGEMYNIF	PSVLDWIPGP	HKRLFRNFGG

	G helix >		< H >		<
	240	250	260	270	280
cyp2a1	LEDFMIEKVR	QNHSTLDPN.	SPRNFIDSFL	IRMQEE.KNG	NSEFHMKNLV
cyp2a4	LEDFITKKVE	HNQRTLDPN.	SPRDFIDSFL	IRMLEEKNP	NTEFYMKNLV
cyp2a5	LEDFITKKVE	HNQRTLDPN.	SPRDFIDSFL	IRMLEEKNP	NTEFYMKNLV
cyp2a6	LEDFAKKVE	HNQRTLDPN.	SPRDFIDSFL	IRMQEEKNP	NTEFYMKNLV
cyp2b1	ILDYIGHIVE	KHRATLDPS.	APRDFIDTYL	LRMEKEKSNH	HTEFHENLM
cyp2b4	INTFIGQSVE	KHRATLDPS.	NPRDFIDVYL	LRMEKDKSDP	SSEFHHQNLI
cyp2b6	INAYIGHSVE	KHRETLDPS.	APKDLIDTYL	LHMEKEKSNA	HSEFHHQNLI
cyp2c2	VKNYITEQIK	EHQKSLDIN.	NPRDFIDCFI	IKMEQEKCNQ	QSEFTIENLL
cyp2c3	QIKFILEKVQ	EHQESLDSN.	NPRDFVDHFL	IKMEKEKHKK	QSEFTMDNLI
cyp2c4	TKNFIMEKVK	EHQKLLDVN.	NPRDFIDCFI	IKMEKE...N	NLEFTLGSIV
cyp2c5	IKNFIMEKVK	EHEKLLDVN.	NPRDFIDCFI	IKMEKE...N	NLEFTLESIV
cyp2c8	TRSYIREKVK	EHQASLDVN.	NPRDFIDCFI	IKMEQEKDNQ	KSEFTIENLV
cyp2c9	MKSYILEKVK	EHQESMDMN.	NPRDFIDCFI	IKMEKEKHNQ	PSEFTIESLE
cyp2c19	MESDILEKVK	EHQESMDIN.	NPRDFIDCFI	IKMEKEKQNG	QNEFTIENLV
cyp2d1	FMALLDNLIA	ENRRTWDPAQ	PPRNLDAFL	AEVEKAKGNP	ESSFNENLR
cyp2d6	FLTQLDELLT	EHRMTWDPAQ	PPRDLTEAFL	AEVEKAKGNP	ESSFNENLR
cyp2e1r	IKEYTLARVK	EHHKSLDPS.	CPRDFIDSLL	IEMEKDKHST	EPLYTLENIA
cyp2e1h	VKEYVSEVRK	EHHQSLDPN.	CPRDLTDCLL	VEVEKEKHS	ERLYTMDGIT
cyp2f1	LRDLIAHSVH	DHQASLDPN.	SPRDFIQCFI	TKMAEEKEDP	LSHFHMDTLL
cyp2f2	MKDLIARSVR	EHQDSLDPN.	SPRDFIDCFI	TKMAEQEKQDF	LSHFHMDTLL

Fig. 1 cont.

	SRS4	I helix	>	<	J helix	>
	290	300	310	320	330	
cyp2a1	MTTSLSFFAG	SETVSSTLRY	GFLLLMKHPD	VEAKVHEEIE	QVIGRNRQPO	
cyp2a4	LTTLNLFFAG	TETVSTTLRY	GFLLLMKYPD	IEAKVHEEID	RVIGRNRQPK	
cyp2a5	LTTLNLFFAG	TETVSTTLRY	GFLLLMKHPD	IEAKVHEEID	RVIGRNRQPK	
cyp2a6	MTTLNLFFAG	TETVSTTLRY	GFLLLMKHPE	VEAKVHEEID	RVIGKNRQPK	
cyp2b1	ISELSLFFAG	TETTSSTTLRY	GFLMLLKYPH	VAEKVQKEID	QVIGSHRLPT	
cyp2b4	LTTLNLFFAG	TETVSTTLRY	GFLMLLKYPH	VTERVQKEIE	QVIGSHRPPA	
cyp2b6	LNTLSLFFAG	TETVSTTLRY	GFLMLLKYPH	VAERVYREIE	QVIGPHRPPE	
cyp2c2	TTVSDVFMAG	TETVSTTLRY	GLLLLMKHPE	VIAKVQEEIE	RVIGRHRSPC	
cyp2c3	TTIWDVFSAG	TDTTSNTLKF	ALLLLLKHPE	ITAKVQEEIE	HVIGRHRSPC	
cyp2c4	IAVFDLFGAG	TETVSTTLRY	SLLLLLKHPE	VAARVQEEIE	RVIGRHRSPC	
cyp2c5	IAVSDLFGAG	TETVSTTLRY	SLLLLLKHPE	VAARVQEEIE	RVIGRHRSPC	
cyp2c8	CTVADLFGAG	TETVSTTLRY	GLLLLLKHPE	VTAKVQEEID	HVIGRHRSPC	
cyp2c9	NTAVDLFGAG	TETVSTTLRY	ALLLLLKHPE	VTAKVQEEIE	RVIGRNRSPC	
cyp2c19	ITAADLLGAG	TETVSTTLRY	ALLLLLKHPE	VTAKVQEEIE	RVIGRNRSPC	
cyp2d1	MVAVDLFTAG	MVTATTTLTW	ALLMLILYD	VQRRVQEEID	EVIGQVRCPE	
cyp2d6	IVVADLFSAG	MVTSTTLAW	GLLMLILHPD	VQRRVQEEID	DVIGQVRRPE	
cyp2elr	VTVADMFFAG	TETVSTTLRY	GLLLLLKHPE	IEEKLHEEID	RVIGPSRMP	
cyp2elh	VTVADLFFAG	TETVSTTLRY	GLLILMKYPE	IEEKLHEEID	RVIGPSRIPA	
cyp2f1	MTTHNLLFGG	TKTVSTTLHH	AFLALMKYPK	VQARVQEEID	LVVGRARLPA	
cyp2f2	MTTHNLLFGG	TETVGTTLRH	AFLILMKYPK	VQARVQEEID	RVVGRSRMPT	

	< J' >	<	K helix	>	SRS5	1-4	1-3
	340	350	360	370	380		
cyp2a1	YEDHMKMPYT	QAVINEIQRF	SNLAPLGIPR	RIIKNTTFRG	FFLPKATDVF		
cyp2a4	YEDRMKMPYT	EAVIHEIQRF	ADLIPMGLAR	RVTKDTKFRD	FLLPKGTEVF		
cyp2a5	YEDRMKMPYT	EAVIHEIQRF	ADLIPMGLAR	RVTKDTKFRD	FLLPKGTEVF		
cyp2a6	FEDRAKMPYM	EAVIHEIQRF	GDVIFMSLAR	RVKKDTKFRD	FLLPKGTEVF		
cyp2b1	LDDRSKMPYT	DAVIHEIQRF	SDLVFIGVPH	RVTKDTMFRG	YLLPKNTEVY		
cyp2b4	LDDRAKMPYT	DAVIHEIQRL	GDLIFFGVPH	TVTKDTQFRG	YVIPKNTEVF		
cyp2b6	LHRAKMPYT	EAVIYEIQRF	SDLEPMGVPH	IVTQHTSFRG	YIIPKDTEVF		
cyp2c2	MQDRSRMPYT	DATVHEIQRY	INLIPNNVPH	TTICNLKFRN	YLIPKGTDLV		
cyp2c3	SQDRSRMPYT	DAVMHEIQRY	VDLVPTSLPH	AVTQDIEFNG	YLIPKGTDLI		
cyp2c4	MQDRSHMPYT	DAVIHEIQRF	IDLEPTNLPH	AVTRDVKFRN	YFIPKGTDLI		
cyp2c5	MQDRSRMPYT	DAVIHEIQRF	IDLEPTNLPH	AVTRDVRFRN	YFIPKGTDLI		
cyp2c8	MQDRSHMPYT	DAVVHEIQRY	SDLVETGVPH	AVTTDTKFRN	YLIPKGTTIM		
cyp2c9	MQDRSHMPYT	DAVVHEVQRY	IDLEPTSLPH	AVTCDIKFRN	YLIPKGTTLI		
cyp2c19	MQDRGHMPYT	DAVVHEVQRY	IDLEPTSLPH	AVTCDVKFRN	YLIPKGTTLI		
cyp2d1	MTDQAHMPYT	NAVIHEVQRF	GDIAPLNLPH	FTSCDIEVQG	FVIPKGTTLI		
cyp2d6	MGDQAHMPYT	TAVIHEVQRF	GDIVPLGVTH	MTSRDIEVQG	FRIIPKGTTLI		
cyp2elr	VRDRVQMPYM	DAVVHEIQRF	IDLVPSNLPH	EATRDITTFQ	YVIPKGTTVI		
cyp2elh	IKDRQEMPYM	DAVVHEIQRF	ITLVPSNLPH	EATRDITFRG	YLIPKGTVVV		
cyp2f1	LKDRAMPYT	DAVIHEVQRF	ADLIPMNLPH	RVTRDTAFRG	FLIPKGTDLI		
cyp2f2	LEDRTSMPYT	DAVIHEVQRF	ADVIPMNLPH	RVTRDTPFRG	FLIPKGTDLI		

Fig. 1 cont.

	< K' > < K'' >						
	390	400	410	420	430		
cyp2a1	PILGSLMTDP	KFFSPSPKDFD	PQNFLDDKGQ	LKKNAAFLEPF	STGKRFLCLGD		
cyp2a4	PMLGSVLKDP	KFFSNPKDFN	PKHFLDDKGQ	FKKSDAFVFP	SIGKRYCFGE		
cyp2a5	PMLGSVLKDP	KFFSNPKDFN	PKHFLDDKGQ	FKKNDAFVFP	SIGKRYCFGE		
cyp2a6	PMLGSVLKDP	SFFSNPQDFN	PQHFLNEKGQ	FKKSDAFVFP	SIGKRNCFGE		
cyp2b1	PILSSALHDP	QYFDHPDSFN	PEHFLDANGA	LKKSEAFMPF	STGKRICLGE		
cyp2b4	PVLSSALHDP	RYFETPNTFN	PGHFLDANGA	LKKNEGFMPF	SLGKRICLGE		
cyp2b6	LILSTALHDP	HYFEKPDADF	PDHFLDANGA	LKKTEAFIFP	SLGKRICLGE		
cyp2c2	TSLSVVLHDD	KEFENPDREF	PGHFLDASGN	FRKSDYFMPF	STGKRVCVGE		
cyp2c3	PSLTSVLYDD	KEFPNPEKFD	PGHFLDESNG	FKKSDYFMPF	STGKRACVGE		
cyp2c4	TSLTSVLHDE	KAFNPVKVFD	PGHFLDESNG	FKKSDYFMPF	SAGKRMCVGE		
cyp2c5	TSLTSVLHDE	KAFNPVKVFD	PGHFLDESNG	FKKSDYFMPF	SAGKRMCVGE		
cyp2c8	ALLTSVLHDD	KEFPNPNIFD	PGHFLDKNGN	FKKSDYFMPF	SAGKRICAGE		
cyp2c9	ISLTSVLHDN	KEFPNPEMFD	PHHFLDEGGN	FKKSKYFMPF	SAGKRICVGE		
cyp2c19	TSLTSVLHDN	KEFPNPEMFD	PRHFLDEGGN	FKKSNYFMPF	SAGKRICVGE		
cyp2d1	INLSSVLKDE	TVWEKPHRFH	PEHFLDAQGN	FVKHEAFMPF	SAGRRACLGE		
cyp2d6	TNLSSVLKDE	AVWEKPFRRH	PEHFLDAQGH	FVKPEAFLEPF	SAGRRACLGE		
cyp2e1r	PTLDSLLYDK	QEFDPDEKFK	PEHFLNEEGK	FKYSDYFKFP	SAGKRVCVGE		
cyp2e1h	PTLDSVLYDN	QEFDPDEKFK	PEHFLNENGK	FKYSDYFKFP	STGKRVCAGE		
cyp2f1	TLNLTVHYDP	SQFLTPEQEFN	PEHFLDANQS	FKKSPAFMPF	SAGRRLCLGE		
cyp2f2	TLNLTVHYDS	DQFKTPQEFN	PEHFLDDNHS	FKKSPAFMPF	SAGRRLCLGE		

	L helix	> 3-1		SRS6	3-2	
	440	450	460	470	480	
cyp2a1	GLARMELFLL	LTTILQNFRE	KFPMKLEDIN	ESPKPLGETRI	IPKYTMSFMPI	
cyp2a4	GLARMELFLF	LTNIMQNFHF	KSTQAPQDID	VSPRLVGFVTI	PPTYTMSFLSR	
cyp2a5	GLARMELFLF	LTNIMQNFHF	KSTQAPQDID	VSPRLVGFVTI	PPTYTMSFLSR	
cyp2a6	GLARMELFLF	FTTVMQNFRL	KSSQSPKID	VSPKLVGFVTI	PRNYTMSFLPR	
cyp2b1	GIARNELFLF	FTTILQNFVS	SSHLAPKID	LTPKESGIGIKI	PPTYQICFSAR	
cyp2b4	GIARTELFLF	FTTILQNFVS	ASPVPPEDID	LTPRESGVGNV	PPSYQIRFLAR	
cyp2b6	GIARAEFLFL	FTTILQNFVS	ASPVAPEDID	LTPQECGVGKI	PPTYQIRFLPR	
cyp2c2	ALARMELFLF	LTAILQNFTE	KPLVNPNNVD	ENPESGIVRV	PPLYRVSFIPV	
cyp2c3	GLARMELFLL	LTTILQHFTE	KPLVDPKID	PTPEVNGEVS	VPPSYELCFVPV	
cyp2c4	GLARMELFLF	LTSILQNFKL	QSLVEPKDLD	ITAVNGEVS	VPPSYQLCFIPI	
cyp2c5	GLARMELFLF	LTSILQNFKL	QSLVEPKDLD	ITAVNGEVS	VPPSYQLCFIPI	
cyp2c8	GLARMELFLF	LTTILQNFNL	KSVDDLKLN	ITAVTKGIVSL	PPSYQICFPIV	
cyp2c9	ALAGMELFLF	LTSILQNFNL	KSLVDPKNLD	ITPEVNGEVS	VPPSYQLCFIPI	
cyp2c19	GLARMELFLF	LTFILQNFNL	KSLIDPKDLD	ITPEVNGEVS	VPPSYQLCFIPI	
cyp2d1	PLARMELFLF	FTCLLQRFSE	.SVPVQGPRP	STHGFAFPPV	APLPYQLCAVVREQGL	
cyp2d6	PLARMELFLF	FTSILLQHFSE	.SVPTGQPRP	SHHGVTAFIV	SPSPYELCAVPR	
cyp2e1r	GLARMELFLL	LSAILQHFNL	KPLVDPEDID	LRNITVGFGRV	PPRYKLCVIPRS	
cyp2e1h	GLARMELFLL	LCAILQHFNL	KPLVDPKID	LSPIHIGEGCI	PPRYKLCVIPRS	
cyp2f1	LLARMELFLY	LTAILQSFSL	QPLGAPEDID	LTPPLSSGGLN	LPRPFQLCLRPR	
cyp2f2	PLARMELFIY	FTSILQNFTE	QPLVDPEDID	LTPPLSSGGLN	LPRPFQLCMHIR	

Fig. 1 cont.

minimization of the raw structure using molecular mechanics. After production of a low energy geometry of CYP2F2, the model was probed using selective substrates of the enzyme, including naphthalene, 3-methylindole, 4-nitrophenol, 4-(methylnitrosamino)-1-(3-pyridyl)-1-butanone (NNK) and ipomeanol. Modelling was assisted by the presence of bound substrate (palmitoleic acid) in the CYP102 crystal structure, and interactive docking procedures (e.g. Dock and FlexiDock) were employed at this stage of the investigation. Following the generation of a structural template using an overlay, these substrate molecules, the candidate compounds 2-isopropyl-naphthalene and 2-(2-hexadecyl)naphthalene, were employed as substrate probes of the CYP2F2 model to ascertain their likely selectivity for the enzyme and subsequent route of metabolism on the basis of analogy with known substrates of CYP2F enzymes. All molecular modelling procedures (including homology modelling, energy minimization and substrate docking) were carried out using the Sybyl software package (Tripos Associates, St. Louis, MO) implemented on a Silicon Graphics Indigo² IMPACT 10000 graphics workstation operating under UNIX.

RESULTS AND DISCUSSION

Figure 2 indicates how the selective substrate naphthalene is able to fit within the putative active site of CYP2F2 such that 1,2-epoxidation can occur. This figure also shows how the putative active site region of CYP2F2 contains residues which interact with the bound substrate, naphthalene, thus orientating it for 1,2-epoxidation. The substrate molecule is held in position via a number of π - π stacking interactions with aromatic amino acid residues within the haem environment. The spatial arrangement of these aromatic side-chains cooperatively assist in orientating the naphthalene molecule such that the 1,2-bond lies directly about the haem iron for oxygenation to occur at that site. Amino acid residues are numbered according to the alignment with CYP102 as presented in Figure 1, which also contains information on substrate recognition site (SRS) regions. In particular, it is the combination of π - π stacking interactions with two phenylalanine residues, Phe181 and Phe78 in the alignment, which orientates the substrate for 1R,2S-oxidation in accordance with known experimental observations /4,6-8/.

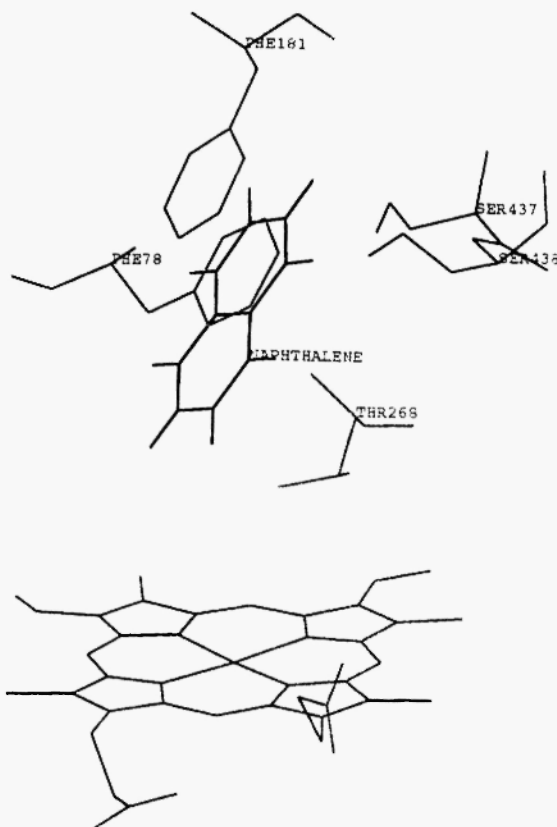


Fig. 2: Naphthalene fitted within the putative active site of CYP2F2. Amino acid residues are numbered according to the alignment with CYP102 shown in Figure 1.

Another selective substrate for CYP2F2 is 3-methylindole, and Figure 3 indicates a likely mode of interaction within the putative active site. In this case, it is a combination of hydrogen bonding to Ser438 (although this only occurs upon entry to the active site) and π - π stacking with Phe181 which orientates the substrate for oxygenation at the 3-methyl position, once again this being in line with the observed findings.

Hydrogen bonding at the same site (i.e. Ser438 in the alignment) also occurs for NNK, 4-nitrophenol and ipomeanol, as shown in

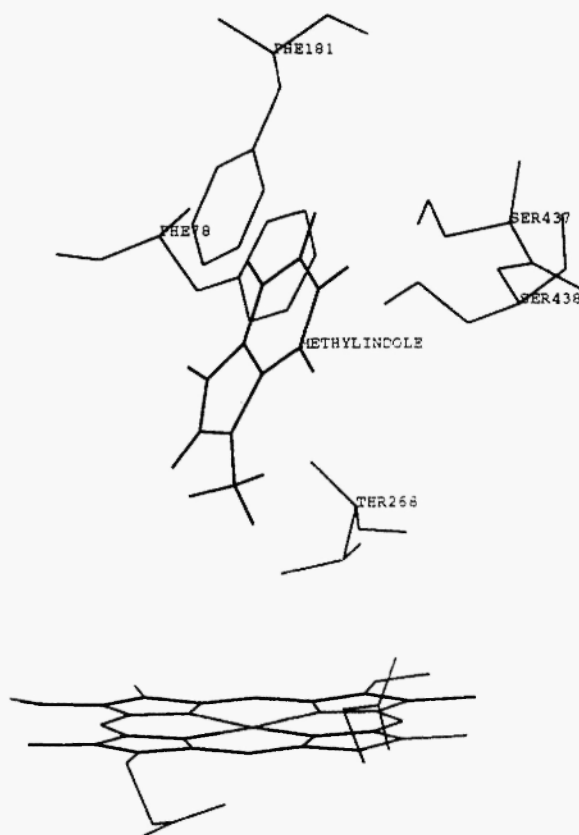


Fig. 3: The active site region of CYP2F2 containing the docked substrate 3-methylindole. Residues are numbered according to the CYP102 alignment (Figure 1).

Figure 4 (where Ser437 is also involved in the latter example) to give the expected metabolites in each case, and this is due to specific active site orientations brought about by amino acid residues in contact with the relevant substrates. Furthermore, the selective inhibitor 5-phenyl-1-pentyne is able to occupy the CYP2F2 site by π - π stacking with Phe181 and Phe78 such that the alkyne group is situated directly above the haem iron (see Fig. 4). This figure includes an overlay of the substrates mentioned above, including naphthalene, 3-methylindole,

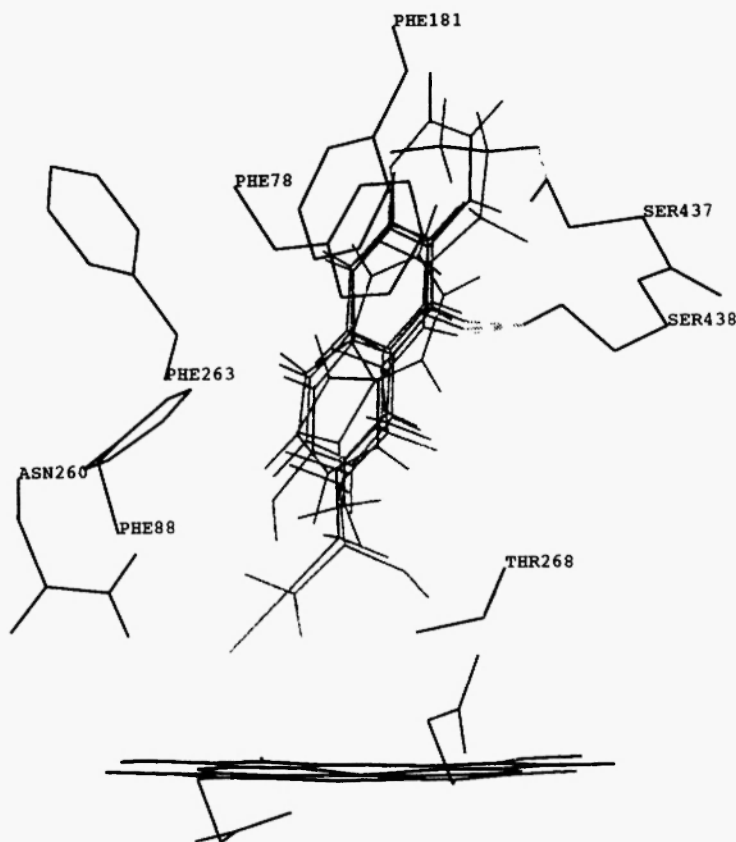


Fig. 4: An overlay of naphthalene, 3-methylindole, 2-isopropyl-naphthalene, 5-phenyl-1-pentyne, 4-nitrophenol, NNK and ipomeanol orientated within the putative active site of CYP2F2 for metabolism at the experimentally observed positions. Hydrogen-bonded interactions are displayed as dashed lines, and residues are numbered according to the alignment with CYP102.

NNK, 4-nitrophenol, ipomeanol and 2-isopropyl-naphthalene, together with the inhibitor 5-phenyl-1-pentyne.

However, in the case of 2-isopropyl-naphthalene, it is unlikely that ring epoxidation occurs and, instead, oxidation of the isopropyl group is more favourable, as shown in the orientation presented in Figure 5,

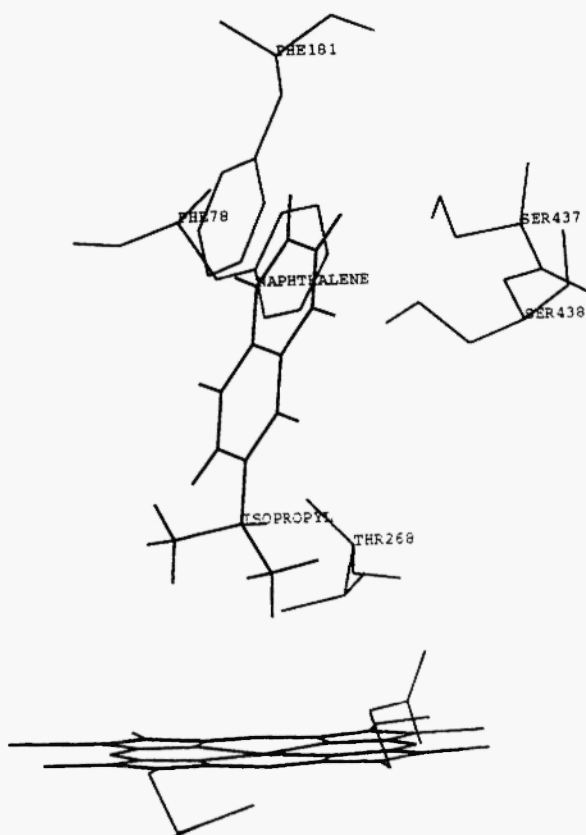


Fig. 5: A potential CYP2F2 substrate, 2-isopropynaphthalene, fitted within the putative active site of this enzyme.

and this is also consistent with information from metabolism studies in the rat [12-14] in which side-chain hydroxylated products are observed. For 2-(2-hexadecyl)naphthalene, the situation is less clear but the presence of the bulky alkyl group would appear to preclude interaction with CYP2F2 and, moreover, the log P value is considerably higher than those of other substrates (see Table 1). Consequently, the model presented in this study indicates that this compound is probably not metabolized via CYP2F2 involvement although this possibility cannot be entirely ruled out.

TABLE 1: CYP2F substrates^a

Compound	log P	pKa	log D _{7.4}	M _r	log M _r	K _m (μM)	ΔG _{bind}
1. Ipomeanol	1.30 ^c	neutral	1.30	168.21	2.2259	N/A	N/A
2. Toluene	2.73	neutral	2.73	92.14	1.9644	N/A	N/A
3. Naphthalene	3.37	neutral	3.37	128.17	2.1078	18 ^d	-6.7302
4. 3-Methylindole	2.60	neutral	2.60	131.18	2.1179	341 ^e	-4.9200
5. 4-Nitrophenol	1.91	7.14	1.91	139.12	2.1434	N/A	N/A
6. 2-Naphthylamine	2.40	4.58 ^b	2.26	143.19	2.1559	N/A	N/A
7. 1-Nitronaphthalene	3.19	neutral	3.19	173.17	2.2385	N/A	N/A
8. 7-Ethoxycoumarin	2.30	neutral	2.30	190.21	2.2792	N/A	N/A
9. 7-Propoxycoumarin	2.86	neutral	2.86	204.24	2.3101	N/A	N/A
10. 7-Ethoxycoumarin	3.61 ^c	9.85 ^b	1.19	241.26	2.3825	N/A	N/A
11. 7-Penloxyresorufin	5.14 ^c	10.06 ^b	2.50	283.35	2.4523	N/A	N/A
12. 7-Benzoyloxyresorufin	4.75 ^c	10.22 ^b	1.98	303.33	2.4819	N/A	N/A

^a Chemicals are either CYP2F1 (human) or CYP2F2 (mouse) substrates or both.
^b Basic (the effect of basicity lowers the range of ionization - corrected log P value), i.e. log D_{7.4}, indicating a relatively narrow window of substrate lipophilicity.
^c Calculated value (Pallas Software, CompuDrug Limited Budapest).
^d Value quoted for CYP2F1 enzyme which is the human orthologue.
^e Value quoted for CYP2F3 enzyme which is a goat orthologue. N/A = data not available.
 $\Delta G_{\text{bind}} = RT \ln K_m$ where K_m is the apparent Michaelis constant.
The pK_a values for alkoxycoumarins were calculated using the Pallas Software.
N/A is also a substrate of this enzyme subfamily.
References: 1, 2, 5, 6, 8, 9, 16/.

TABLE 2
Estimation of enzyme-substrate binding energies (kcal.mol⁻¹)

Substrate	Naphthalene	3-Methylindole
log P _{expt}	3.37	1.91
ΔG _{part}	-4.780	-2.709
ΔG _{π-π}	-2.70 (3π-π interactions)	-0.9 (1 π-π interaction)
ΔG _{hbond}	0	-2.0 (1 hydrogen bond)
ΔG _{rotors}	0	+0.6 (1 rotatable bond)
ΔG _{bind} ^{calc}	-7.480	-5.009
K _m (μM)	4.7	340 [#]
ΔG _{bind} ^{expt}	-7.557	-4.920

log P_{expt} = experimental log P value, where P is the octanol/water partition coefficient.

ΔG_{bind}^{calc} = ΔG_{part} + ΔG_{π-π} + ΔG_{hbond} + ΔG_{rotors}. The use of this relation and the average values of the following components has been reported recently /15/.

ΔG_{part} = RTlnP where P is the octanol/water partition coefficient, R is the gas constant and T is the absolute temperature (310°K).

ΔG_{π-π} = -0.9 kcal.mol⁻¹ for each π-π stacking interaction.

ΔG_{hbond} = -2.0 kcal.mol⁻¹ for each hydrogen bond interaction.

ΔG_{rotors} = +0.6 kcal.mol⁻¹ for each rotatable bond in the molecule.

[#] Data for CYP2F3, the goat orthologue of CYP2F2.

References for K_m data: /6,8/.

From the published K_m values of naphthalene and 3-methylindole, the binding energy (ΔG_{bind}) can be estimated* using the relationship ΔG = RTlnK (see Table 2) and it is possible to calculate values for binding affinity from the modelled interactions (see Figs. 2 and 3)

* Assuming that the dissociation constant, KD, is approximately equal to Km, the Michaelis constant for the reaction catalyzed by CYP2F.

based on the average energies for π - π stacking and hydrogen-bonded interactions, together with the partitioning free energy (calculated from experimental log P values) as shown in Table 2. The data presented in Table 2 show that it is possible to obtain good estimates of the experimental binding affinities for these two substrates via a linear combination of the factors contributing to their overall ΔG_{bind} . It is also interesting that this relatively straightforward procedure gives reasonably good results for the calculation of substrate binding affinity, and similar findings have been observed for other CYP2 family substrates /15/.

CONCLUSIONS

The results of homology modelling of the mouse P450, CYP2F2, are consistent with substrate metabolism data in showing that selective substrates are able to fit the putative active site for metabolism at the experimentally observed positions. Orientation for metabolism is achieved via complementary contacts with a relatively small number of key amino acid residues which enter into π - π stacking and/or hydrogen-bonded interactions with typical CYP2F2 substrates, such as naphthalene, ipomeanol, NNK, 3-methylindole and 3-nitrophenol. The selective inhibitor, 5-phenyl-1-pentyne, also fits within the active site such that the alkynic group is likely to become activated and subsequently bind to the haem moiety (i.e. act as a suicide substrate). Alkyl naphthalenes, such as 2-isopropyl naphthalene, may also interact with CYP2F2 but will probably become metabolized at the alkyl group rather than via epoxidation of the aromatic ring system. However, 2-(2-hexadecyl)naphthalene is unlikely to be a CYP2F2 substrate due to the bulky aliphatic side-chain. Consequently, homology modelling of CYP2F2 enables one to evaluate likely substrate selectivity and route of metabolism for this subfamily.

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